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Synthesis of styrenes through the decarboxylation of *trans*-cinnamic acids by plant cell cultures

Masumi Takemoto * and Kazuo Achiwa

School of Pharmaceutical Sciences, University of Shizuoka, 52-1 Yada, Shizuoka 422-8526, Japan

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Abstract

A new method has been developed for the synthesis of styrenes through the decarboxylation of *trans*-cinnamic acids by plant cell cultures at room temperature. 4-Hydroxy-3-methoxystyrene (**2a**), 3-nitrostyrene (**2d**) and furan (**2e**) were synthesized quantitatively. © 1999 Elsevier Science Ltd. All rights reserved.

Keywords: decarboxylation; enzymes and enzyme reactions; plants; carboxylic acids and derivatives.

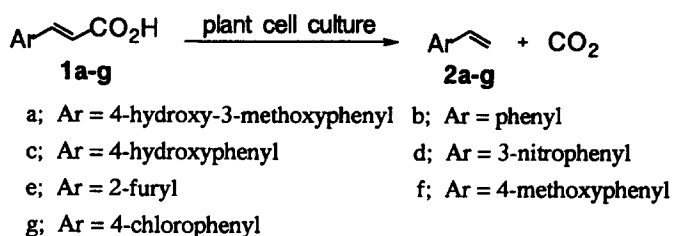
Decarboxylation of *trans*-cinnamic acids is the most widely used chemical method for preparing styrenes or stilbenes. A typical decarboxylation is carried out by heating under reflux at 200–300°C for 4–5 h in quinoline in the presence of a Cu powder (Y>50%). Quinoline is useful as a solvent for the decarboxylation of unsaturated acids because it is basic enough to form the required carboxylate anion and also because it boils at a temperature favorable for decarboxylation. This method, however, needs a high temperature.

On the other hand, the known decarboxylative enzymes are mainly as follows: (i) pyruvate decarboxylase;¹ (ii) oxalate decarboxylase;² (iii) glutamate decarboxylase;³ (iv) benzoylformate decarboxylase;⁴ (v) aconitate decarboxylase;⁵ and (vi) aspartate 4-decarboxylase.⁶

In the case of *trans*-cinnamic acids, β -phenylacrylic acid was decarboxylated by *Aspergillus niger* to give styrene.⁷ *Aerobacter* has been found to decarboxylate *trans*-4-hydroxycinnamic acid to the corresponding 4-hydroxystyrene.⁸ However, only a few attempts for the decarboxylation of other *trans*-cinnamic acids by a decarboxylase have been reported.

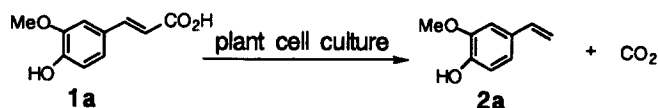
Here we report a novel method for decarboxylation of *trans*-cinnamic acids (**1a–g**) by plant cell cultures to the corresponding styrenes or furan (**2a–g**). This decarboxylation takes advantage of the mild reaction conditions for preparing styrenes.

* Corresponding author.



When *trans*-ferulic acid (**1a**) was subjected to plant cell culture in a medium, 4-hydroxy-3-methoxystyrene (**2a**)⁹ was given quantitatively as shown in Table 1. The decarboxylative reaction was performed by two methods, that is: (A) with freely suspended plant cells in the stationary phase after 10 days of incubation (10 g of cells in 20 ml of a medium); and (B) with homogenized plant cell culture in 0.1 M phosphate buffer solution (pH 6.4). In the case of *Catharanthus roseus*, **1a** was quantitatively decarboxylated to **2a** not only with method (A), but also with (B) (entries 1, 2). But in the cases of *Nicotiana tabacum* and *Daucus carota*, the decarboxylation proceeded with only method (B) (entries 4, 6).

Table 1
Decarboxylation of *trans*-ferulic acid (**1a**) with plant cell culture



Entry	Plant cell culture	Method	Time (day)	Product 2a C.Y. (%)	Recovery 1a C.Y. (%)
1	<i>C. roseus</i>	A	5	quant.	0
2	<i>C. roseus</i>	B	3	quant.	0
3	<i>N. tabacum</i>	A	5	trace	84
4	<i>N. tabacum</i>	B	5	quant.	0
5	<i>D. carota</i>	A	5	0	100
6	<i>D. carota</i>	B	5	30	65
7	<i>C. sinensis</i>	A	5	0	100
8	<i>C. sinensis</i>	B	5	0	100

Next we tried the decarboxylation of other *trans*-cinnamic acids (**1b–g**) using method (B) as shown in Table 2. The decarboxylation of **1d** and **1e** with *Camellia sinensis* gave 3-nitrostyrene **2d** and furan **2e** quantitatively (entries 11, 12). In the case of **1c**, **2c** was given in 30–32% yield by *C. roseus* or *D. carota* (entries 7, 9). In the case of **1b**, **1f** and **1g**, the corresponding products **2b**, **2f** and **2g** were given in low chemical yields (entries 4, 13, 16). These styrenes (**2b**,¹⁰ **2d**,¹¹ **2f**¹¹ and **2g**¹¹) and furan **2e**¹² were chemically synthesized by the decarboxylation of *trans*-cinnamic acids (**2b**, **2d–g**) in the presence of a copper powder in quinoline at 185–195°C for 2–4 h (Y>50%). A major advantage of our method is that the decarboxylation with plant cell culture proceeds mildly at room temperature. Studies are now in progress to shorten the reaction time.

For a typical experiment, we used suspension-cultured cells which had originally been isolated from *C. roseus*, *N. tabacum* 'Bright Yellow-2', *D. carota*, and *C. sinensis* as described in our previous papers.^{13–15}

Table 2
Decarboxylation of cinnamic acids (**1a-g**) with plant cell cultures

$\text{Ar}-\text{CH}=\text{CH}-\text{CO}_2\text{H} \xrightarrow[\text{5 days}]{\text{plant cell culture}} \text{Ar}-\text{CH}=\text{CH}_2 + \text{CO}_2$				
Entry	Substrate	Plant cell culture	Product 2a-g C.Y. (%)	Recovery 2a-g C.Y. (%)
1	1a	<i>C. roseus</i>	quant.	0
2		<i>N. tabacum</i>	quant.	0
3		<i>D. carota</i>	30	65
4	1b	<i>C. roseus</i>	10	80
5		<i>N. tabacum</i>	trace	86
6		<i>D. carota</i>	trace	85
7	1c	<i>C. roseus</i>	30	63
8		<i>N. tabacum</i>	5	84
9		<i>D. carota</i> (root ^{a)})	32	55
10		<i>C. sinensis</i>	trace	83
11	1d	<i>C. sinensis</i>	quant.	0
12	1e	<i>C. sinensis</i>	quant.	0
13	1f	<i>C. sinensis</i>	10	76
14		<i>D. carota</i> (seed ^{b)})	trace	85
15		<i>D. carota</i> (root)	trace	87
16	1g	<i>D. carota</i> (seed)	10	78
17		<i>C. sinensis</i>	trace	82

a) *D. carota* cell line derived from root.

b) *D. carota* cells derived from seedling.

In the case of method (A), a substrate (50 mg) was added to the freely suspended *C. roseus* (B-5 medium pH 5.5), *N. tabacum* 'Bright Yellow-2' (MS medium, pH 5.8), *D. carota* (MS medium, pH 5.8), and *C. sinensis* (B-5 medium, pH 5.8). The mixture was shaken at 25°C on a rotary shaker (110 rpm) in the dark. At the conclusion of the reaction, the incubation mixture was filtered, the filtered cells were washed with CH₂Cl₂, and the filtrates were combined. The combined mixture was extracted with CH₂Cl₂. The organic layer was dried over anhydrous MgSO₄ and subjected to column chromatography. In the case of method (B), 10 g of plant cells were homogenized in 10 ml 0.1 M phosphate buffer (pH 6.4). A substrate (50 mg) was added to the homogenate. The subsequent procedure was the same as for method (A).

Acknowledgements

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References

1. Juni, E. *J. Biol. Chem.* **1961**, *236*, 2302.
2. Simazono, H.; Hayaishi, O. *J. Biol. Chem.* **1957**, *227*, 151.
3. Shukuya, R.; Schwert, G. W. *J. Biol. Chem.* **1960**, *235*, 1649.
4. Gunsalus, C. F.; Stanier, R. Y.; Gunsalus, I. C. *J. Bact.* **1953**, *66*, 548.
5. Bentley, R.; Thiessen, C. P. *J. Biol. Chem.* **1957**, *226*, 703.
6. Novogrodsky, A.; Meister, A. *J. Biol. Chem.* **1964**, *239*, 879.
7. Jaminet, F. *J. Pharmac. Berg.* **1950**, *5*, 191.
8. Finkle, B. J.; Lewis, J. C.; Corse, J. W.; Lundin, R. E. *J. Biol. Chem.* **1962**, *237*, 2926–2931.
9. The data for **2a** ($\text{Ar-CH}_x=\text{CH}_A\text{H}_B$) is as follows; $^1\text{H NMR}$ (CDCl_3) ppm: 3.92 (3H, s, OCH_3), 5.11 (1H, d, HB, $J_{\text{BX}}=8.9$ Hz), 5.58 (1H, d, HA, $J_{\text{AX}}=17.8$ Hz), 6.63 (1H, q, HX), 6.81–6.95 (3H, m, Ph). $^{13}\text{C NMR}$ (CDCl_3) ppm: 55.87 (OCH_3), 108.03 (Ph), 111.43 ($=\text{CH}_2$), 114.34 (Ph), 120.05 (Ph), 130.26 (Ph), 136.62 ($-\text{CH}=\text{}$), 145.62 (Ph), 146.58 (Ph).
10. Abbott, T. W.; Johnson, J. R.; Clarke, H. T.; Brethen, M. R. *Org. Synth. Coll.* Vol. I, 440–442.
11. Wiley, H.; Smith, N. R.; Arnord, R. T.; Parham, W. E.; Davis, D. D. *Org. Synth. Coll.* Vol. IV, 731–734.
12. Wilson, W. C.; Adams, R.; Gauerke, C. G. *Org. Synth. Coll.* Vol. I, 274–275.
13. Takemoto, M.; Achiwa, K.; Stoykov, N.; Chen, D.; Kutney, J. P. *Phytochemistry* **49**, 423–426.
14. Takemoto, M.; Moriyasu, Y.; Achiwa, K. *Chem. Pharm. Bull.* **1995**, *43*, 1458–1461.
15. Takemoto, M.; Yamamoto, Y.; Achiwa, K. *Chem. Pharm. Bull.* **1998**, *46*, 419–422.